## AMENDMENTS TO THE CLAIMS

1. (Withdrawn) An ozone gas measurement method comprising the steps of:
preparing a sensing element in which a dye that changes in a light absorption
characteristic of a visible region upon reaction with ozone gas is deposited in a pore of a porous
material;

exposing the sensing element to a measurement environment for a predetermined time; and

measuring an ozone gas amount in a measurement target gas on the basis of a change in the dye before and after exposing the sensing element to the measurement environment for a predetermined time.

- 2. (Withdrawn) A method according to claim 1, wherein the step of measuring the ozone gas amount comprises the step of measuring a change in light transmittance.
- 3. (Withdrawn) A method according to claim 2, wherein the step of measuring the ozone gas amount comprises the steps of

measuring the light transmittance of the sensing element to obtain a first transmittance, and

measuring an ozone gas amount in the measurement target gas on the basis of the first transmittance, and a second transmittance before the sensing element measured in advance is exposed to the measurement environment for the predetermined time.

- 4. (Withdrawn) A method according to claim 1, wherein at least some pores in the porous material are coupled to pores on a surface of the porous material.
- 5. (Withdrawn) A method according to claim 1, wherein a pore in the porous material has such a pore diameter as to attain a predetermined transmittance in the visible light region.
- 6. (Withdrawn) A method according to claim 5, wherein the pore diameter is not more than 20 nm at which the dye can enter the pore.
- 7. (Withdrawn) A method according to claim 1, wherein the dye comprises an aromatic

compound having a diazo group.

- 8. (Withdrawn) A method according to claim 7, wherein the aromatic compound comprises one material selected from the group consisting of benzene, naphthalene, and anthracene.
- 9. (Withdrawn) A method according to claim 7, wherein the dye comprises a compound having any one of a hydroxyl group, a sulfurous acid group, and primary to tertiary amino groups.
- 10. (Withdrawn) A method according to claim 1, wherein the dye comprises a triphenylmethane stain.
- 11. (Withdrawn) A method according to claim 1, wherein the dye contains fuchsonimine.
- 12. (Withdrawn) A method according to claim 1, wherein the dye contains indigo.
- 13. (Withdrawn) A method according to any one of claims 10 and 11, wherein the sensing element further comprises a material having an alkali characteristic in addition to the dye.
- 14. (Withdrawn) A method according to claim 1, wherein the sensing element further comprises an acid gas sorbent in addition to the dye.
- 15. (Withdrawn) A method according to claim 14, wherein the acid gas sorbent comprises one material selected from the group consisting of glycerol and triethanolamine.
- 16. (Withdrawn) A method according to any one of claims 7 and 12, wherein the sensing element further comprises an acid in addition to the dye.
- 17. (Withdrawn) A method according to claim 16, wherein the acid comprises one acid selected from the group consisting of hydrochloric acid, acetic acid, sulfuric acid, and phosphoric acid.
- 18. (Withdrawn) A method according to claim 16, wherein the sensing element further comprises a hygroscopic compound in addition to the dye and the acid.

- 19. (Withdrawn) A method according to claim 18, wherein the hygroscopic compound comprises one material selected from the group consisting of glycerol and ethylene glycol.
- 20. (Withdrawn) A method according to any one of claims 7 and 12, wherein the sensing element further comprises a buffer in addition to the dye.
- 21. (Currently Amended) An ozone gas sensing element comprising:
- a transparent porous material comprising a water film formed in a pore of the porous material; and
- a dye which is deposited in a the pore of said porous material and changes in a light absorption characteristic of a visible region upon reaction with ozone gas.
- 22. (Original) An element according to claim 21, wherein at least some pores in said porous material are coupled to pores on a surface of said porous material.
- 23. (Original) An element according to claim 21, wherein a pore in said porous material has such a pore diameter as to attain a predetermined transmittance in the visible light region.
- 24. (Original) An element according to claim 23, wherein the pore diameter is not more than 20 nm at which the dye can enter the pore.
- 25. (Previously Presented) An ozone gas sensing element comprising: a transparent porous material; and
- a dye which is deposited in a pore of said porous material and changes in a light absorption characteristic of a visible region upon reaction with ozone gas, wherein the dye comprises an aromatic compound having a diazo group.
- 26. (Original) An element according to claim 25, wherein the aromatic compound comprises one material selected from the group consisting of benzene, naphthalene, and anthracene.
- 27. (Original) An element according to claim 25, wherein the dye comprises a compound having any one of a hydroxyl group, a sulfurous acid group, and primary to tertiary amino groups.

- 28. (Original) An element according to claim 21, wherein the dye comprises a triphenylmethane stain.
- 29. (Previously Presented) An ozone gas sensing element comprising:
  a transparent porous material; and
  a dye which is deposited in a pore of said porous material and changes in a light
  absorption characteristic of a visible region upon reaction with ozone gas, wherein the dye

contains fuchsonimine.

- 30. (Currently Amended) An element according to claim 21, wherein the dye contains indigo carmine.
- 31. (Currently Amended) An element according to claim 29, wherein said sensing element further comprises a material having an alkali characteristic in addition to the dye.
- 32. (Previously Presented) An ozone gas sensing element comprising: a transparent porous material; and
- a dye which is deposited in a pore of said porous material and changes in a light absorption characteristic of a visible region upon reaction with ozone gas, wherein said sensing element further comprises an acid gas sorbent in addition to the dye.
- 33. (Original) An element according to claim 32, wherein the acid gas sorbent comprises one material selected from the group consisting of glycerol and triethanolamine.
- 34. (Previously Presented) An element according to claim 30, wherein said sensing element further comprises an acid in addition to the dye.
- 35. (Original) An element according to claim 34, wherein the acid comprises one acid selected from the group consisting of hydrochloric acid, acetic acid, sulfuric acid, and phosphoric acid.
- 36. (Original) An element according to claim 34, wherein said sensing element further comprises a hygroscopic compound in addition to the dye and the acid.

- 37. (Original) An element according to claim 36, wherein the hygroscopic compound comprises one material selected from the group consisting of glycerol and ethylene glycol.
- 38. (Previously Presented) An element according to claim 30, wherein said sensing element further comprises a buffer in addition to the dye.
- 39. (Original) An element according to claim 38, wherein the buffer comprises phosphoric acid and sodiumdihydrogenphosphate dehydrate.
- 40. (Withdrawn) An ozone gas measurement apparatus comprising:
  - a light-emitting unit;
  - a light-detecting unit;
  - a sensing element; and
  - a signal processing unit,

wherein said light-emitting unit emits light having a predetermined wavelength, said sensing element is interposed between said light-detecting unit and said light-receiving unit, and comprises a porous material, and a dye which is deposited in a pore of the porous material and changes in a light absorption characteristic of a visible region upon reaction with ozone gas,

said light-detecting unit comprises a light-receiving surface arranged to face said light-emitting unit, receives, via said sensing element, light emitted by said light-emitting unit, and outputs a signal corresponding to a light quantity received by the light-receiving surface, and

said signal processing unit calculates an ozone gas amount on the basis of the signal output from said light-detecting unit and a light absorption characteristic, obtained in advance, of said sensing element which contains the dye before reaction with the ozone gas.

- 41. (Withdrawn) An apparatus according to claim 40, wherein at least some pores in the porous material are coupled to pores on a surface of the porous material.
- 42. (Withdrawn) An apparatus according to claim 40, wherein a pore in the porous material has such a pore diameter as to attain a predetermined transmittance in the visible light region.
- 43. (Withdrawn) An apparatus according to claim 42, wherein the pore diameter is not more than 20 nm at which the dye can enter the pore.

- 44. (Withdrawn) An apparatus according to claim 40, wherein the dye comprises an aromatic compound having a diazo group.
- 45. (Withdrawn) An apparatus according to claim 44, wherein the aromatic compound comprises one material selected from the group consisting of benzene, naphthalene, and anthracene.
- 46. (Withdrawn) An apparatus according to claim 44, wherein the dye comprises a compound having any one of a hydroxyl group, a sulfurous acid group, and primary to tertiary amino groups.
- 47. (Withdrawn) An apparatus according to claim 40, wherein the dye comprises a triphenylmethane stain.
- 48. (Withdrawn) An apparatus according to claim 40, wherein the dye contains fuchsonimine.
- 49. (Withdrawn) An apparatus according to claim 40, wherein the dye contains indigo.
- 50. (Withdrawn) An apparatus according to any one of claims 47 and 48, wherein said sensing element further comprises a material having an alkali characteristic in addition to the dye.
- 51. (Withdrawn) An apparatus according to claim 40, wherein said sensing element further comprises an acid gas sorbent in addition to the dye.
- 52. (Withdrawn) An apparatus according to claim 51, wherein the acid gas sorbent comprises one material selected from the group consisting of glycerol and triethanolamine.
- 53. (Withdrawn) An apparatus according to any one of claims 44 and 49, wherein said sensing element further comprises an acid in addition to the dye.
- 54. (Withdrawn) An apparatus according to claim 53, wherein the acid comprises one acid selected from the group consisting of hydrochloric acid, acetic acid, sulfuric acid, and phosphoric acid.

- 55. (Withdrawn) An apparatus according to claim 53, wherein said sensing element further comprises a hygroscopic compound in addition to the dye and the acid.
- 56. (Withdrawn) An apparatus according to claim 55, wherein the hygroscopic compound comprises one material selected from the group consisting of glycerol and ethylene glycol.
- 57. (Withdrawn) An apparatus according to any one of claims 44 and 49, wherein said sensing element further comprises a buffer in addition to the dye.
- 58. (Withdrawn) An apparatus according to claim 57, wherein the buffer comprises phosphoric acid and sodiumdihydrogenphosphate dehydrate.
- 59. (Previously Presented) An element according to claim 28, wherein said sensing element further comprises a material having an alkali characteristic in addition to the dye.
- 60. (Previously Presented) An element according to claim 25, wherein said sensing element further comprises an acid in addition to the dye.
- 61. (Previously Presented) An element according to claim 25, wherein said sensing element further comprises a buffer in addition to the dye.